

Research Article

Nanomaterial's synthesis, types and their use in bioremediation and agriculture

Praveen Kumar*

Department of Biochemistry, COBS&H, CCS Haryana Agricultural University, Hisar-125004, India.

Accepted 15 September, 2021

Diseases, pests and pollutants are the major factors of agricultural production. They reduce a large proportion of agricultural production by reducing plant growth and development. Nano science can be utilized in the remediation of these problems in agriculture, as it provides a safer, efficient and effective way of agrochemicals delivery to the plants, resulting in increased agricultural production. Nanoparticles are widely used in the agricultural sector because of their distinctive properties. Studies have shown the influence of nanoparticles on plant growth and production. Nanoparticles act as effective carriers in the delivery of agrochemicals to plants. They provide site targeted delivery of nutrients and thus, prevent wastage of nutrients applied for plant growth and productivity. Bioremediation of pollutants is an emerging technology that provides Bio-Nano Materials for the protection of agriculture from pollution. This review aims to present and focus on the latest techniques used for the reduction of environmental pollution and improved agricultural production. This review speculates about the biosynthesis of nanomaterials from different sources like plants, fungi and bacteria along with chemical and organic synthesis from carbon, silver and gold. The role of nanoscience in detecting plant diseases and removal of heavy metals. Application of Nanoscience in storing, production, processing and transport of agricultural materials. It is also emphasized that Nanoscience may transform agriculture through innovation of new techniques like Precision farming, improvement of plants to engross nutrients, targeted use of inputs, detection and control of diseases and withstand environmental pressures. Further, efforts have been made in describing that nanoparticles may act as a better substitute for agricultural plant's growth and nutrition improvement by lowering the content of pollutants and pre-detection of diseases in plants. The biosynthetic route of nanomaterial synthesis could emerge as a better and safer option for environmental pollution reduction. Thus, nanoscience may increase agricultural production to feed a huge population in near future.

Key words: Bioremediation; nanomaterials; biosynthetic; sustainable; ecofriendly

INTRODUCTION

The challenging task of the 21st-50th century is to clean the environment in an eco-friendly and sustainable way, so, that the agricultural production can be increased. Although it is a new area still it is rapidly growing. Microbes are the emerging Nano-factories and have potentials for environmental clean-up methods. Most of the biogenic nanomaterials yielded satisfactory results. They can increase the production in an agricultural system. India is an agricultural dependent country. Its economy is also dependent on agricultural production. All animals, directly or indirectly, depends on agricultural production, for their food supply. Thus any change in agricultural production

may affect the development and economy of any country like India. Diseases, lack of nutrients and pollution are three major reasons for the reduction in agricultural production. Among diseases, microbial diseases are becoming a vast problem in agriculture and the chemicals used in the remediation of these diseases are producing problems in animals that are consuming them. These toxic chemicals (*Fungicide, Herbicide and Bactericide*) causes a major loss of animal species like honey bee and birds which are beneficial for agricultural production (Wiyaratn, et al. 2012). These toxic chemicals also cause loss of beneficial bacteria's from soil which is involved in the decomposition of various waste materials and thus increasing soil fertility (Mandal, et al. 2006). Nowadays farmers adding nutrients more than their requirement by the

*Corresponding author. Praveen Kumar, Email: praveenkholahau.ac.in.

plants. This extra quantity of nutrients are becoming wastage and directly affecting the farmer's income from agricultural production. Among pollutants, heavy metals are becoming a major problem in agricultural production. Heavy metals cause harmful effects to the agricultural plants and reduce their production. Heavy metals enter animals through food sources from plants and cause toxicity in them also. In short, if we are gaining on one hand but we are losing more on the other hand. So, as an outcome, we are losing a lot. This loss of agricultural production can be prevented by employing nanoparticle science technology for providing nutrients, diseases protection and protection from heavy metals pollution. Nanoscience or nanotechnology is a branch of science dealing with studies involving nano size (10⁻⁹ size of meter) particles. Nano can be applied to any unit of measure for example you can consider nanograms and nanoliters. The concept of nanoscience was first time driven by Richard Feynman in 1959 and from then it was growing very fast in all areas of research and development worldwide (Jain, et al. 2010). Nowadays, nanoscience is the subject of extensive research in all areas (Li, et al. 2007). It may bring the next industrial as well as the green revolution in the world (Satishkumar, et al. 2009). Nanoscience provides many environmental benefits to agriculture. These may be divided into nano-fertilizer, disease remediation and reduction of pollutants like heavy metals. The specific nanoscience application for onsite remediation of soil, air, groundwater and wastewater treatment are also in development. The size of nanoparticles has significance in their physiological and biochemical properties. It behaves completely different from its main part. The reduction in the size changes redox properties, thermodynamic properties and internal cohesive forces of the particle or material. It causes easy delivery and interaction of the particle with the active site (Li, et al. 2007). Nanoparticle synthesis using microorganisms comes under green nanoscience technology. It may provide a better alternative for diseases control in plants (Hsu, et al. 2015). Nanoscience is a persistent solution for our environment. It is needed to utilize new methods like nanoparticle science for better development in agriculture. Nanoscience can help us to design new materials with peculiar properties and reproducibility for agricultural development (Ndlovu, et al. 2020). In this concern, the scientific community should continuously focus on a very challenging and relevant research's direction, which is the development of nanoparticles capable of heavy metal remediation from soil and water. The concept of nanoscience may help us to understand the mechanism of transformation of heavy metals into plants and animals, more accurately at the molecular level. The synthesis of materials at the nanosize dimension will facilitate us to design new useful systems for heavy metal removal from plants. This review will provide information on different systems and methods which may be utilized to overcome heavy metal toxicity among agricultural lands across the globe.

NANOPARTICLES

The particle of very small size (<100 nm), of any material, is called a nanoparticle of that particular element. They are atomic or molecular aggregates of 1-100 nm dimension and their physicochemical properties modify drastically as compared to the bulk material. The properties of nanoparticles depend and

may vary with their size i.e. the size of nanoparticles define their properties. Nanoparticles are more mobile and more reactive. They are roughly divided into organic and inorganic. The first category of organic nanoparticles comprises carbon nanoparticles. The second category inorganic nanomaterials comprise magnetic nanomaterials, noble metal nanoparticles like silver, gold nanoparticle and semiconductor nanoparticles like titanium dioxide, zinc oxide. Nanomaterials are also produced naturally by volcanic and lunar eruptions (Li, et al. 2013). These are known as natural nanoparticles but there are other classes of nanoparticles that are produced by anthropogenic activities like the exhaust of diesel, coal combustion and fumes of welding works; these are known as incidental nanoparticles. The nanoparticles synthesized by man are called engineered nanoparticles. The man-made or engineered nanoparticles includes nano zinc, nano aluminium, TiO₂, ZnO, quantum dots etc (Hočevár, et al. 2013). The smaller size of nanoparticles makes them more useful for use as sensors that can be easily deployed in remote locations. Currently, nanomaterials are being proved successful both in efficiency as well as cost-effective and also in the environmental friendly way in utilizing them as an alternative for current treatment materials for agricultural remediation (Rao, et al. 2007; Rodríguez-León, et al. 2013). The use of biologically synthesized nanomaterials has been increasing in the field of agriculture and medicine because of their stable, eco-friendly and cost-effective nature (Popescu, et al. 2010). They should be utilized in wider applications. The biological synthesis of nanomaterials is more advantageous over other methods due to ease of fast synthesis, ease of toxicity control, ease of size control, cost-effective and environmentally friendly methodology (Umashankari, et al. 2012). Nanoscience is being utilized extensively for the protection of agriculture from bacterial disease and pollutants including chemicals like heavy metals (Petla, et al. 2012).

UNIQUE PROPERTIES OF NANOPARTICLES AND THEIR CONTRIBUTION TO PLANT PRODUCTION AND GROWTH

The properties of nanoparticle changes with its size. Nanoparticles show different and unique properties when compared with their bulked product. Some examples are sown here below-Table 1. These properties showed a way to encapsulate and slow the release of agrochemicals used in plant growth and protection. The major roles of nanoparticles in increasing agricultural production are summarized in Table 2.

Table 1. Properties of nanoparticles along with their bulked forms.

Materials	Nanoparticle	Bulked
Copper	Hard	Soft
Gold	Chemically active	Chemically inactive
Silicon	Conductor	Insulator
Titanium dioxide	Colourless	White

Table 2. Contribution of nanoscience in plant production and growth.

Sl. No.	Contribution of Nanoscience
1.	Nano Materials (NM) can be effectively used in plant germination and growth.

2. The carbon nanotubes can be used as regulators of seed germination and plant growth.
3. Multi-Wall Carbon Nanotubes (MWCNTs) can enhance the growth of tobacco cell culture by 55-64% when compared to control at a wide range of concentrations from 5-500 $\mu\text{g ml}^{-1}$.
4. At low concentrations, activated carbon enhanced cell growth.
5. Nanotechnology provides controlled and efficient use of fertilizers, pesticides and other agrochemicals for better growth and production of plants (Gudkov SV, et al. 2020).
6. Chitosan nanoparticles are proved more effective anti-fungal agents against the pathogen (*Fusarium solani*).
7. Silver nanoparticles act as antimicrobial agents and protect plants from microbial diseases.
8. Copper nanoparticles may act as bio-pesticides and protect plants from pests.
9. Zinc based nanoparticles inhibit fungal growth by damaging conidiophores.

SYNTHESIS OF NANOPARTICLES

Traditionally, there are several methods for generating nanoparticles, including gas condensation, abrasion, chemical precipitation, ion embedding, pyrolysis, and hydrothermal production. In abrasion, macro- or micro-scale elements are crushed in a ball mill, an earthly ball mill, or other size-reducing apparatus. The resulting particles are air categorized to recover nanoparticles. In pyrolysis, a vaporous precursor (liquid or gas) is forced through an orifice at high pressure and burned. The resulting solid (a version of soot) is air categorized to recover oxide particles from by-product gases. Old-style pyrolysis often results in aggregates and agglomerates rather than single primary particles. Ultrasonic nozzle Spray Pyrolysis (USP) on the other hand aids in preventing agglomerates from forming. These methods are costly for nanoparticle synthesis, there are other methods to produce nanoparticles cheaply as:

BIOGENIC PRODUCTION OF VARIOUS NANOPARTICLES

The microorganisms and plant extracts are used generally for a cheaper synthesis of nanoparticles (Figure 1). The main reaction involved in the biosynthesis of the nanoparticle is the oxidation/reduction reaction. It is a bottom-up approach for the synthesis of the nanoparticle. The biochemical antioxidants in the plant sample and biochemical enzymes in the microbial sample are responsible for the reduction of metal into their nanoparticles. Currently, there is million tons of production of nanoparticles worldwide and is probable to increase intensely in the near upcoming (Sekhon, 2010). The nanomaterials are materials having a particular dimension between 1-100 nm, so along with vascular plants; the microorganisms like bacteria, yeasts, algae, fungi, and actinomycetes may be used for nanoparticles biosynthesis (Saxena, et al. 2014).

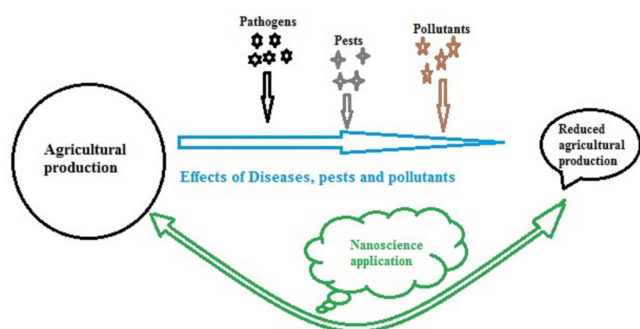


Figure 1. Effects of diseases, pests and pollutants.

NANOPARTICLES PRODUCED BY PLANTS

Nanoparticle synthesis from plant extract is called a green synthesis of the nanoparticle (Shebl, et al. 2019, Hassan, et al. 2020). Currently, it is gaining prominence because of the single-step involved in biosynthesis. So, it is a time-saving process along with no toxicants and the presence of natural capping agents (Saxena, et al. 2014). Plants material is easily available, they are safe to handle and possess a large variability of metabolites that aid in reduction. All these reasons make plants more advantageous over others for nanoparticle synthesis. Currently, there are many plants are being investigated for their importance in nanoparticles synthesis (Table 3). Phytochemicals consume less time for the reduction of metal ions, compared to fungi and bacteria. It shows plants materials as a better candidate for the biosynthesis of nanomaterials, compared to bacteria and fungi. By utilizing plant tissue culture and downstream processing techniques, one can synthesis nanomaterials of metallic and oxide at an industrial scale if the metabolic status of the plants are addressed appropriately. It is found that the nature of nanoparticles produced from this method varies with the plant to plant, mode of application, size, and quantity (Mathew, et al. 2010). It is revealed by the scientist that the investigation on nanoscience basically for plants is in its early stages; more depth study is required to reveal about biochemical, physiological and molecular mechanisms of plants in concern to nanomaterials and more work is required to explore the mechanism of nanomaterials actions, their behaviour towards biomolecules and their effect on gene regulation and expressions in plants. Generally, plant-based nanoparticles involve three mechanisms viz. adsorption, reduction and desorption mechanisms in the heavy metal removal process from plants, soil and water (Figure 2).

Table 3. Plants used for the synthesis of nanoparticles.

Nanopar- ticles	Plants species	References
Gold and silver nanoparti- cles	<i>Citrus sinensis</i>	Siddiqui MH, et al. 2015
	<i>Diospyros kaki (Persim- mon)</i>	Sathyavathi R, et al. 2010
	<i>Pelargonium graveolens</i>	Sweeney RY, et al. 2004
	<i>Hibiscus rosa sinensis</i>	Pope JH, et al. 2000
	<i>Coriandrum sativum</i>	Parida UK, et al. 2011
	<i>Embllica officinalis</i>	Arfin T, et al. 2013, Red- dy MC, et al. 2015
	<i>Phyllanthium</i>	Ankamwar B, et al. 2005
Gold nanoparti- cles	<i>Mushroom extract</i>	Kavitha KS, et al. 2013
	<i>Terminalia catappa</i>	Philip D, et al. 2009
	<i>Banana peel</i>	Ankamwar B, 2010
	<i>Mucuna pruriens</i>	Bankar A, et al. 2010
	<i>Cinnamomum zeylani- cum</i>	Arulkumar S, et al. 2010
	<i>Medicago sativa</i>	Jung JY, et al. 2011
	<i>Magnolia kobus and</i>	Blaney LM, et al. 2007,
	<i>Dyopiros kaki</i>	Reddy MC, et al.2015
	<i>Allium cepa L.</i>	Kathad U, et al. 2014
	<i>Azadirachta indica A.</i>	Pavani KV, et al. 2012
	<i>Juss.</i>	Parashar V, et al. 2009
	<i>Camellia sinensis L.</i>	Boruah SK, et al. 2012
	<i>Chenopodium album L.</i>	Dwivedi AD, et al. 2011
	<i>Justicia gendarussa L.</i>	Fazaludeena MF, et al. 2017
	<i>Macrotyloma uniflorum</i>	Aromal SA, et al. 2012
	<i>(Lam) Verde</i>	Wiyaratn W, et al. 2012
	<i>Mentha piperita L.</i>	MubarakAli D, et al.
	<i>Mirabilis jalapa L.</i>	2012
	<i>Syzygium aromaticum (L)</i>	Philip D, 2009
	<i>Terminalia catappa L.</i>	Raghunandan D, et al.
	<i>Amaranthus spinosus</i>	2010

Sili-con-Germanium (Si-Ge) nanoparticles	<i>Freshwater diatom Stauroneis sp.</i>	Yadav KK, et al. 2017, Roychoudhury A, 2020
Silver nanoparticles	<i>Elettaria cardamomom</i> <i>Parthenium hysterophorus</i> <i>Ocimum sp.</i> <i>Euphorbia hirta</i> , <i>Nerium indicum</i> <i>Azadirachta indica</i> <i>Brassica juncea</i> <i>Pongamia pinnata</i> <i>Clerodendrum inerme</i> <i>Gliricidia sepium</i> <i>Desmodium triflorum</i> <i>Opuntia ficus indica</i> <i>Coriandrum sativum</i> <i>Carica papaya (fruit)</i> <i>Pelargoneum graveolens</i> <i>Aloe vera extract</i> <i>Capsicum annum</i> <i>Avicennia marina</i> <i>Rhizophora mucronata</i> <i>Cerriops tagal</i> <i>Rumex hymenosepalus</i> <i>Pterocarpus santalinus</i> <i>Sonchus asper</i>	Fazaludeena MF, et al. 2017 Gurunathan S, et al. 2009, Vankar PS, et al. 2010 Mandal D, et al. 2006 Gericke M, et al. 2006 SweeneyRY, et al.2004, Pugazhenthiran N, et al. 2009, Pavani KV, et al. 2012 Arfin T, et al. 2013 Vahabi K, et al. 2011 Raut Rajesh W, et al. 2009 Farooqui MA, et al. 2010 Monica RC, et al. 2009 Ahmad N, et al. 2011 Sekhon BS, 2010 Babu MG, et al. 2009 Jain D, et al. 2009 Jain PK, et al. 2006 Chandran SP, et al. 2006 Lin D, et al. 2007 Jain D, et al. 2010 Gnanajobitha G, et al. 2012 Rodríguez-León, et al. 2013 Dhas SP, et al. 2013 Kasthuri J, et al. 2009
Palladium nanoparticles	<i>Cinnamomum zeylanicum Blume.</i> <i>Cinnamomum camphora L.</i> <i>Gardenia jasminoides Ellis.</i> <i>Soybean (Glycine Max) L.</i>	Gudkov SV, et al. 2020 Pavani, KV, et al. 2012 Tang WW, et al. 2012 Jiang S, et al. 2009
Magnetic Nanoparticles	<i>Aloe vera</i>	Philip D, 2010
Lead Nanoparticles	<i>Vitis vinifera L.</i> <i>Jatropha curcas L.</i>	Petla RK, et al. 2012 Priya MM, et al. 2011
Nanoparticles of silver, nickel, cobalt, zinc and copper	<i>Brassica juncea</i> , <i>Medicago sativa</i> and <i>Helianthus annuus</i>	Juibari MM, et al. 2011
Platinum nanoparticles	<i>Diopyros kaki</i> <i>Ocimum sanctum L.</i>	Jung JY, et al. 2011 Sathyavathi R, et al. 2010

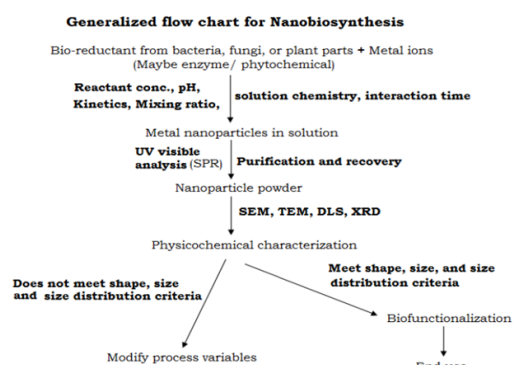


Figure 2. General outline for the biosynthesis of nanoparticles.

NANOPARTICLES PRODUCED BY BACTERIA

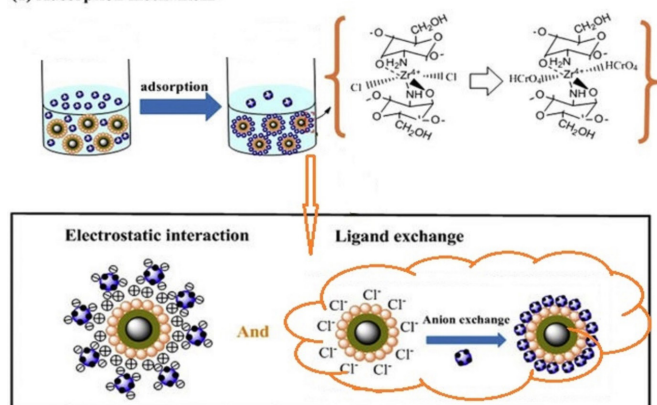
Bacteria possess the peculiar property of mobilization and immobilization of elements and in some cases, they may precipitate metals even in nanometer size. Therefore, Bacteria are called bio-factory for the production of nanomaterials like silver, gold, palladium, titanium, magnetite, cadmium and platinum. It is a new method of nanoparticle synthesis. In this method, bacterial enzymes are involved in catalyzing a specific breakdown reaction and produce nanoparticles (Bali, et al. 2006). The polysaccharides, vitamins, enzymes, biodegradable polymers and biological systems may be utilized for the synthesis of nanoparticles in the same way. The enzymes which are produced in extracellular secretion possess more advantage of the production of a large number of nanomaterials ranging size between 100-200 nm in a comparably pure form, and free from other materials like cellular proteins. These nanomaterials may be further purified by filtration easily. The metal binding and S-layer properties of bacteria make them more useful in the technical application of bioremediation and nanoscience. The different types of bacteria used for nanomaterials are shown (Table 4). The mechanism involved in silver nanoparticle synthesis by *Pseudomonas aeruginosa* is explained in Figure 3. The quality of nanoparticles produced by these techniques may be controlled by optimization of bacterial growth controlling parameters like enzymatic activities and cellular activities. Thus, there is a lot more to study to make proper use of this technique for nanomaterial synthesis. It is a more appealing field because of an ecofriendly, more stable, safe and less expensive process.

Table 4. Bacteria are used for the synthesis of nanoparticles.

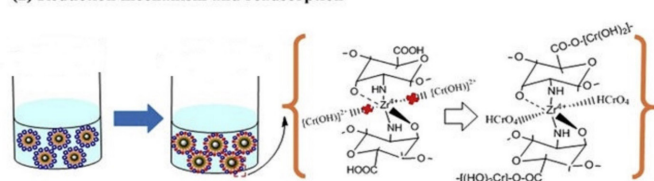
Nanoparticles	Bacterium species	References
ZnS nanoparticles	Sulphate reducing bacteria of the family Desulfobacteriaceae	Sintubin L, et al. 2009, Smitha SL, et al. 2009, Lee JH, et al. 2007, Zhong LS, et al. 2007, Raut RW, et al. 2010, Srivastava NK, et al. 2008
CdS nanoparticles	Clostridium thermoaceticum Klebsiella aerogens Escherichia coli	Olenin AY, et al. 2011 Wang P, et al. 2016 Robinson T, et al. 2001
As-S nanotubes	Shewanella sp.	Li S, et al. 2007
Palladium nanoparticles	Desulfovibrio desulfuricans NCIMB 8307	Nanda A, et al. 2009
Gold nanoparticles	Alkalothermophilic actinomycete Thermomonospora sp. Pseudomonas aeruginosa Lactobacillus strain	Sathishkumar M, et al. 2009 Sathishkumar M, et al. 2009 Ingale AG, et al. 2013 Sintubin L, et al. 2009
Magnetic nanoparticles	Magnetosirillum magneticum Sulphate reducing bacteria	Mathew L, et al. 2010 Gnanadesigan M, et al. 2012

Gold nanowires	Rhodospseudomonas capsulate	Husseiny MI, et al. 2007
Silver nanoparticles	Bacillus cereus	Pope JH, et al. 2000
	Oscillatoria willei	Murugadoss G, et al. 2009
	NTDMO1	Li H, et al. 2011
	Escherichia coli	Klaus-Joerger T, et al. 2001, Priya MM, et al. 2011, Joglekar S, et al. 2011
	Pseudomonis stutzeri	
	Bacillus subtilis	Jampilek J, et al. 2011
	Bacillus sp.	Rathore KS, et al. 2008
	Bacillus cereus	Gardea-Torresdey JL, et al. 2002
	Bacillus thuringiensis	Li L, et al. 2004
	Lactobacillus strains	Lee JH, et al. 2007
	Pseudomonas stutzeri	Klaus-Joerger T, et al. 2001
	Corynebacterium	Monica RC, et al. 2009
	Staphylococcus aureus	Narayanan KB, et al. 2008
	Ureibacillus thermosphaericus	Velammal SP, et al. 2016

(1) Adsorption mechanism



(2) Reduction mechanism and readsorption



(3) Desorption



Figure 3. General mechanisms involved in heavy metal removal by plant-based nanoparticles from plants, soil and water.

NANOPARTICLES PRODUCED BY YEAST AND FUNGI

Fungi produce a larger amount of extracellular enzymes than bacteria, which shows more influence on nanomaterial synthesis. The documented mechanistic aspect governing nanoparticle synthesis by using fungi is explained here. The nanomaterials with well-defined dimensions can be synthesized

by using fungi. Compared to bacteria, fungi can be used for larger production of nanomaterials because fungi secrete a larger amount of proteins which directly increases the formation of nanomaterials (Mathew, et al. 2010). Moreover, the protein isolates from fungi can also be used for nanoparticle formation. The enzymes produced by fungi during metal nanoparticle synthesis reduce salts to their metallic solid nanomaterials by the catalytic effect (Sweeney, et al. 2004). It is major weaknesses of this synthesis method by microorganisms and if need be corrected if this method is to participate with other methods. The other disadvantage of this method is that it generates nanomaterials at a slower rate than detected when plant extract is used. The other thing to be considered for the fungal synthesis of nanomaterials is that fungi should possess certain properties like high production of required metabolites/enzymes, high growth rate, easy handling in large scale production and require low cost for production techniques (Roychoudhury, 2020). Generally, fungi are observed as the best source for nanomaterial synthesis over other biological systems because of their easy handling, cost-effectiveness, and wide diversity. Genetic engineering may be employed for an ecofriendly and improved method of nanoparticle synthesis (Shankar, et al. 2003). In terms of the quality of nanomaterials and utilization of nutrients for microbial growth, yeast strains have assured benefits over bacteria (Popescu, et al. 2010).

ROLE OF NANOSCIENCE IN DETECTING PLANT DISEASES

There is a strong need to detect plant diseases in the early stage so that they can be protected from microbial agents and a great loss of food and agriculture can be prevented. Plant diseases are the most limiting factor for reducing crop productivity; the initial requirement to achieve control of plant disease is to recognize their causal organisms correctly. The identification process by involving traditional methods consumes a lot of time and inputs. But nanotechnological techniques such as nanoparticles and Quantum Dots (QDs), nano-imaging, nanostructured platforms, and nanopore tools for DNA sequencing requires less time and can be used for enhancing crop production and to ensure high-quality monitoring. In recent years, nanotechnology has emerged as a new diagnostic tool. Nano diagnostic kits help in preventing any epidemics at a much early stage which quickly detect the potential of plant pathogens and allows experts to help the farmers, thus nanotechnology can be applied for more cost-effective, quick, and accurate diagnostic plans of plant diseases. Nanotechnology may also be used for the delivery of pesticides at targeted/infected plant tissues, for pathogen detection and control of wound healing in plants. Nanoscience can be utilized in real-time monitoring systems, distributed throughout the fields, to monitor soil conditions and crop conditions, by providing or connecting an autonomous nanosensor to a GPS. This technique reduces the time required for observation of the crop in fields. It provides results within a few hours that are more accurate, simple and portable. This technique does not involve any complicated procedures and is so simple to be utilized by the farmers. The union of nanoscience with sensors technology may create new methods which allow early detection of diseases development and environmental changes.

Nanoscience provides a new way of changing crop management methods and improving plant nutrition. It reduces the problem of leaching, photolytic degradation, hydrolysis, and microbial degradation. It has reduced the wastage of nutrients by providing how a very low amount of required minimal effective concentration, reach the target site of crops.

NANOMATERIALS FOR HEAVY METALS REMOVAL

Heavy metals are distributed among soil, air, and water by natural or human activities like weathering, biodegradation, volcanic eruptions and mining, agrochemical industries; respectively. Heavy metals are increasing with human civilization. Any element having an atomic weight greater than 63.5 and has a specific gravity greater than 5.0 can be called heavy metal (Mueller, et al. 2012). Heavy metals are divided into toxic metals (Hg, Cr, Pb, Zn, Cu, Ni, Cd, Co, Sn); precious metals (Pd, Pt, Ag, Au, Ru) and radionuclides (U, Th, Ra, Am) (Kim, et al. 2013). These metals are non-biodegradable and very difficult to eliminate naturally from soil, air, and water. Generally, all metals are toxic when their values increase from their permissible limits. Heavy metals cause toxic effects in human's body when it exceeds the permissible limits. Various

technologies have been used for heavy metal remediation like precipitation, reverse osmosis, advanced oxidation, coagulation, photoelectrochemistry, and flocculation. Each differs in the degree of remediation efficiency (Hočevár, et al. 2013, Turhan, et al. 2013). Currently, nanoscience has provided a great approach for heavy metal toxicity remediation by providing different nanomaterials. This technique is very beneficial over other techniques, some of the benefits are increased efficiency, reduced consumption and substitution of more abundant and less toxic materials than ones used before. Generally adsorbing nanomaterials have shown great and effective work against heavy metal toxicity removal (Popescu, et al. 2010). Nanoparticles for heavy metal removal are divided into carbon-based nanoparticles (carbon nanotubes and graphene) and inorganic nanoparticles (based on metal oxides and metals). Different combination of nanoparticles is also developed for heavy metal removal. The disposal of nanoparticles may lead to release in the aquatic ecosystem which may prove harmful for them, this should be solved. The plants utilized in heavy metal removal are summarized in (Table 5). The interaction of nanoparticles with plant's biomolecules and removal of heavy metal cations and anions is explained in Figure 4.

Table 5. Plants utilized for nanoparticle generation and their role in bioremediation of heavy metals.

Plant species used for Nanoparticle generation	Best heavy metal bioremediation	Reference
Noaea mucronata	Pb(98%), Zn(79.03%), Cu(73.38%), Cd(72.04%) and Ni(33.61%)	Sekhon B.S, 2010
Euphorbia macroclada	Pb(92%), Zn(76.05%), Cu(74.66%), Cd(69.08%) and Ni(31.50%)	Kalavathy H, 2010

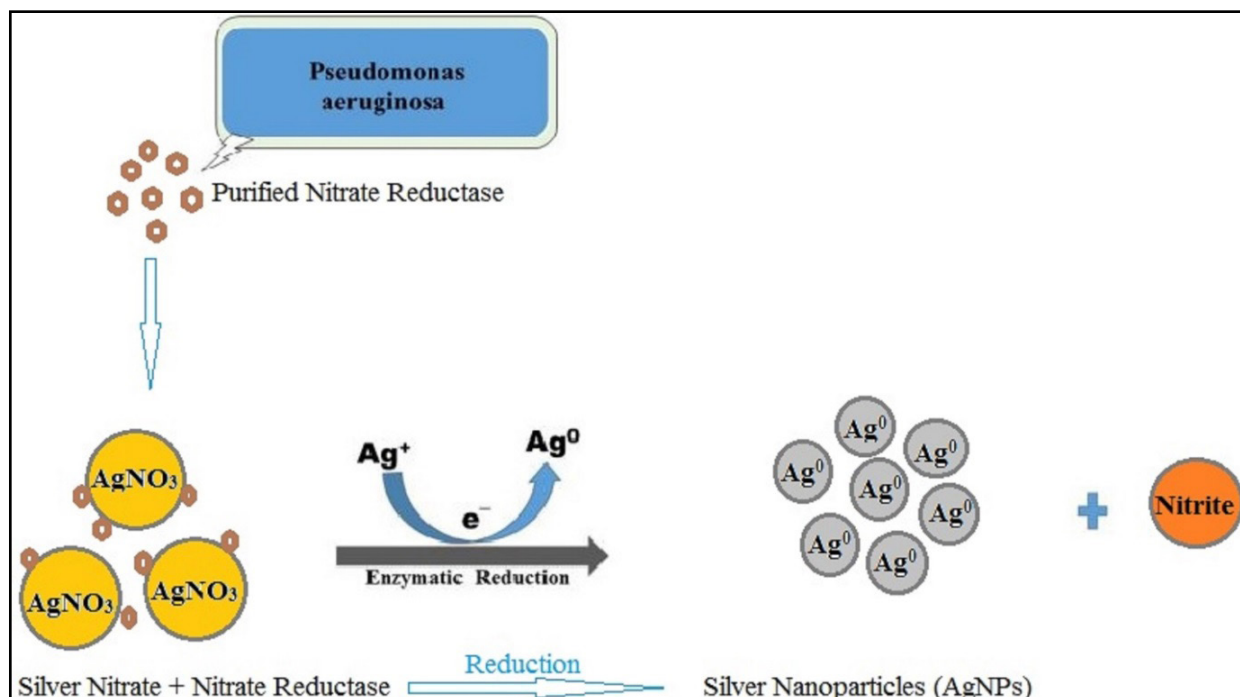


Figure 4. Mechanistic illustration of silver nanoparticle synthesis by bacterial involvement.

Carbon-based nanomaterials

Activated carbon: Activated carbon is produced from freely available carbonaceous precursors such as wood, coal, coconut shells and agricultural wastes. They possess high porosity and a large surface area. They are broadly used in wastewater treatments. These nanomaterials possess a high potential for the removal of organic and inorganic pollutants from contaminated water. Activated carbon has a significant weak acidic ion-exchange character which is involved in the removal of trace metal contaminants from wastewater by adsorption. (Zhong, et al. 2009) in his work observed the adsorption of pentavalent arsenic on granular activated carbon. The activated carbon synthesized from coconut tree sawdust was observed as an adsorbent for Cr (VI) and removes from wastewater (Nazeruddin, et al. 2016). Activated carbon is investigated for adsorption and stability of mercury on it (He, et al. 2008). The adsorption ability of powdered activated carbon was tested against Cu (II) and Pb (II) and was found 0.85, 0.89 $\mu\text{mol g}^{-1}$; respectively. A new sodium polyacrylate grafted activated carbon was synthesized by gamma radiation to enhance functional groups on its surface. This procedure may be applied to other adsorbents to enhance the efficiency of metal ions adsorption by activated carbon (Ewecharoen, et al. 2009). Because of the low cost and high adsorption ability of activated carbons, they are appropriate materials for heavy metal removal.

Carbon nanotubes: Carbon nanotubes are like pillars for nanoscience. They are generally used in nanomaterial synthesis. Carbon nanotubes are divided into two groups as single-walled and multiwall nanotubes (Mallikarjuna, et al. 2012). Their potential as a superior adsorbent for heavy metals has been studied widely (Nair, et al. 2002; Lu, et al. 2006; Ahmaruzzaman, 2011; Zhu, et al. 2016). Currently, scientists are focusing on nanomaterial synthesis which shows specific adsorption towards a single solute in the target material and ignoring the interaction potentials between mixtures that may affect the adsorption process (Rao, et al. 2007, Wang, et al. 2009, Yang, et al. 2010, Hong, et al. 2012, Li, et al. 2007). Carbon nanotubes have been used as a better adsorbent for the removal of heavy metals like Pb^{2+} , Cd^{2+} , Ni^{2+} , and Cu^{2+} (Petla, et al. 2012). The mechanism involved in heavy metal removal by carbon nanotubes comprises electrostatic attraction, adsorption, precipitation and chemical interaction between the metal ion and functional group present on the surface of carbon nanotubes. The drops in pH of the solution, when metal binds to carbon nanotubes indicate the transfer or release of the proton from carbon nanotubes where metal binds to it. Sometimes phenolic groups are also involved in heavy metal binding to carbon nanotubes. Moreover, an increase in metal ion concentration also increases the pH values of the solution and indicates the adsorption of heavy metal (Rao, et al. 2007). Various carbon nanotubes have been synthesized to improve adsorption capability and minimize weaknesses under different environmental circumstances, for the removal of heavy metals. The combination of magnetic properties of iron oxide with adsorption properties of carbon nanotubes may prove rapid, effective and promising technology for hazardous heavy metals removal (Kasthuri, et al. 2009). The maximum adsorption

capacity of different heavy metals on carbon nanotubes differs among different heavy metals.

Graphenes: Graphene-based adsorbents have a great adsorption capacity for removing heavy metal pollutants from water and soil (Park, et al. 2005, Zhang, et al. 2003). Many researchers are focusing on and putting their efforts into investigating the potential application of graphene oxide nanomaterials in eliminating heavy metals from water and soil. The adsorption properties of graphene nanomaterials may be improved if graphene nanomaterial is synthesized by employing the Hummers method which incorporates many oxygen-containing groups like carboxyl and hydroxyl groups on the surface of nanomaterials (Mueller, et al. 2012). These functional groups increase the surface area of graphene oxide nanomaterial which further increases the adsorption capacity of nanomaterials for heavy metals. The mechanism of heavy metal removal by graphene nanomaterials is the same as was discussed for carbon nanotubes. The high dispersion nature of graphene oxide nanomaterials makes it more advantageous over carbon nanotubes. The other nanomaterials that have been studied include carbon fibre, carbon, nanoporous carbon and their complexes (Ahmed, et al. 2018, Prabhu, et al. 2012). They show exceptional adsorbent capacity for heavy metal removal and the rates of removal are persistent with pollution control requirements. Their efficiency depends on well-developed internal pore structures, large specific surface areas, and the presence of functional groups on surface nanomaterials.

Other nanomaterials: Along with carbon-based nanomaterials, other nanomaterials such as iron-based nanomaterials, photocatalytic nanomaterials, silica, alumina, and manganese dioxide nanomaterials are also explored for heavy metal pollutant removal. These nanoparticles have been engrossed for their outstanding adsorption properties. Some other nanomaterials which are utilized in the removal of heavy metal from soil, air and water are TiO_2 -Based Nanomaterials, ZnO-based nanomaterials, Iron oxide nanomaterials, bimetallic nanoparticles, and Metal based-nanomaterials (Pejman, et al. 2015). All possess excellent efficiency for the removal of heavy metal pollution from agricultural lands and irrigation water. But still, a lot of further research is required on nanoscience to remove heavy metals more efficiently and significantly from agricultural lands and water.

Nanotechnology has an enormous role in the field of nano-fertilizer. Nano-based fertilizer can do balanced nutrient supply, stimulated crop growth, improved crop quality, regulate nutrient migration to the environment, reduction of cost of production due to its less requirement per hectare (Mikula, et al. 2020), helps in precision farming, increased nutrient use efficiency of plants, increased drought and salt stress tolerance to plants (Kalia, et al. 2020). The two most important applications of nanotechnology in nano-fertilizers include the controlled release of fertilizers and the bio-availability of nutrients (Muhammad, et al. 2020). Apart from these two roles nanotechnology can also be used for the target modulation and regulation of Plant Growth-Promoting Rhizobacteria (PGPR) at the genetic or proteomic level (Fu, et al. 2020). In India, nano-fertilizers are used mostly to boost zinc uptake in wheat and other cereals,

but there is a gap in research and its application by the farmers. Agriculture is facing various challenges such as climate changes, environmental issues, urbanization, sustainable use of resources and accumulation of pesticide and fertilizers. More research and work is needed to make these solutions possible at ground level i.e. to the farmer's level. There should be subsidized programs on the use of these technologies by the farmers so that maximum benefits of these technologies can be seen on ground level reality. All farmers should be made aware of the use and benefits of these products for their crops and particular farming process. Without these efforts the gap between research and its application by the farmers cannot be filled. Dr. G. Pandey in his review has shown the application and challenges of nanotechnology for sustainable agriculture in India (Pandey, 2020). The impact of different nano-fertilizers on the yield and growth characteristics of particular crops is illustrated in Table 6 and Table 7.

Nanotechnology has a promising approach in controlling the diseases of crops. It is beneficial over conventional approaches, such as reduced input requirements, better efficacy and lower eco-toxicity (Usman, et al. 2020). Nanomaterials may act as antimicrobial agents, biostimulants that induce plant innate immunity, and as carriers for active ingredients such as elicitors, micronutrients, and pesticides in controlling diseases (Rehmanullah, et al. 2020). Thus nanotechnology may provide potential benefits in the field of disease management and agricultural adaptations shortly.

The use of nanomaterials as agrochemicals either as fertilizers or pesticides is scientifically being explored before it could be used in agriculture for a general farm practice. The properties of nanomaterials such as type of material, shape, size, crystal phase, solubility, exposure and dosage concentrations are considered to be of the potential risk to human health (Shukla, 2019). However, some experts point out that food products containing nanomaterials available in the market are probably safe to eat, then also this area needs to be investigated more actively because detailed studies are required to address the impact and safety concern of nanomaterials within the

human body when exposed to nanomaterials *via* food (Guha, et al. 2020).

Researchers have to develop proper assessment approaches to assess the impact of nanomaterials or nano-fertilizers on biotic and abiotic components of the ecosystem (Heinisch, et al. 2019). Amongst the different issues concerning the use of nanomaterials in agriculture, the nature of accumulation of nanomaterials in the environment and the focus on the edible part of plants might be the important issues before their use.

Nevertheless, Dr. S.K. Singh in his review has discussed the impending use of nanotechnology in the fields of food and agriculture such as nano-fertilizer, plant nutrition, and plant protection and suggested that nanoparticles may be toxic to the environment and biological system due to their changed physicochemical properties (Singh, et al. 2020). Thus while considering the benefits of nanomaterials researchers should also focus on the probable risk that may arise due to that nanomaterials.

Role of nanoparticles in precision farming

Advanced biosensors for precision farming are possible only by nanoparticles or nanochips. The modern nanotechnology has the potential to address the problems of conventional agriculture and can revolutionize this sector. The potential uses and benefits are enormous that includes insect pest management *via* formulations of nanomaterial based pesticides and insecticides, increase in agricultural productivity using nanoparticles encapsulated fertilizers for slow and sustained release of nutrients and water. Precision agriculture uses AgTech to collect detailed field and crop data, mainly using sensing technologies combined with geo-referencing. Over the last few decades, many new technologies have been developed for precision farming (Duhan, et al. 2017). Some of these are Satellite Positioning (GPS) system, remote sensing, automated steering system, geo-mapping and variable Rate Technology (VRT). Some other roles of nanoparticles in precision farming are listed in Figures 5 and 6.

Table 6. Impact of nano-fertilizers on yields of different crops under varying climatic conditions.

Nanofertilizers	Crops	Yield increment (%)
Nanofertilizer+urea	Rice	10.2
Nanofertilizer+urea	Rice	8.5
Nanofertilizer+urea	Wheat	6.5
Nanofertilizer+urea	Wheat	7.3
Nano-encapsulated phosphorous	Maize	10.9
Nano-encapsulated phosphorous	Soybean	167
Nano-encapsulated phosphorous	Wheat	28.8
Nano-encapsulated phosphorous	Vegetables	12.0-19.7
Nano chitosan-NPK fertilizers	Wheat	14.6
Nano chitosan	Tomato	20.0
Nano chitosan	Cucumber	9.3
Nano chitosan	Capsicum	11.5
Nano chitosan	Beet-root	8.4
Nano chitosan	Pea	20
Nanopowder of cotton seed and ammonium fertilizer	Sweet potato	16
Aqueous solution on nanoiron	Cereals	8-17
Nanoparticles of ZnO	Cucumber	6.3
Nanoparticles of ZnO	Peanut	4.8
Nanoparticles of ZnO	Cabbage	9.1

Nanoparticles of ZnO	Cauliflower	8.3
Nanoparticles of ZnO	Chickpea	14.9
Rare earth oxides nanoparticles	Vegetables	7-45
Nanosilver+allicin	Cereals	4-8.S
Iron oxide nanoparticles+calcium carbonate nanoparticles + peat	Cereals	14.8-23.1
Sulfur nanoparticles+silicon dioxide nanoparticles+synthetic fertilizer	Cereals	3.4%-45%

Table 7. Impact of different nano-fertilizers on different growth and survival characteristics of particular crops.

Nanofertilizers	Crops	Imparted characteristics
Nanoparticles of ZnO	Chickpea	Increased germination, better root development, higher indoleacetic acid synthesis.
Nano silicon dioxide	Maize	Drought resistance, increment in lateral root roots number along with and shoot length.
Nano silicon dioxide	Maize	Increased leaf chlorophyll
Nano silicon dioxide	Tomato	Taller plants and increased tuber diameter.
Colloidal silica+NPK fertilizers	Tomato	Increased resistance to pathogens.
Nano-TiCl	Spinach	Improved vigor indices and 28% increased chlorophyll.
Polyethylene+indium oxide	Vegetables	Increased sunlight absorption
Polypropylene+indium-tin oxide	Vegetables	Increased sunlight utilization
Gold nanoparticles+sulfur	Grapes	Antioxidants and other human health benefits.
Kaolin+SiO ₂	Vegetables	Improved water retention
Bentonite+N-fixing bacteria inoculation	Legumes	Improved soil fertility and resistance to insect-pest.
Nanocarbon+rare earth metals+N fertilizers	Cereals	Improved nitrogen use efficiency
Stevia extract+nanoparticles of Se+organo-Ca+rare-earth elements + chitosan	Vegetables	Enhanced root networking and root diameter
Nano-iron slag powder	Maize	Reduced incidence of insect-pest
Nano-iron+organic manures	Cotton	Controlled release of nutrients acts as an effective insecticide and improves soil fertility status.

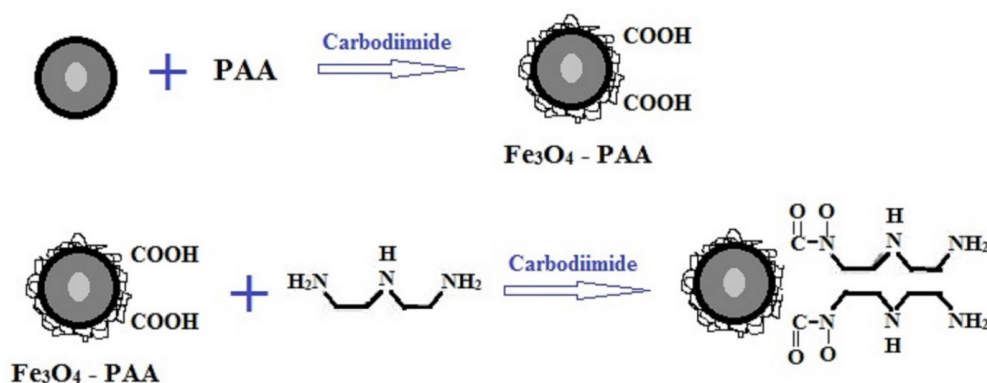


Figure 5. Iron oxide nanoparticle generation and activation of Poly-Acrylic Acid (PAA) by amino-functionalization on iron oxide nanoparticles as a new approach in making magnetic nano-adsorbent for metal cation and anions removal.

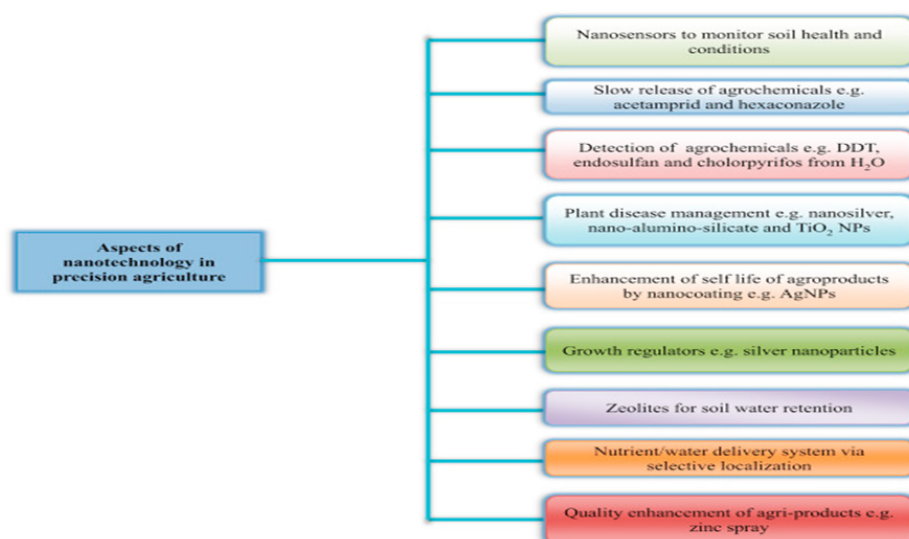


Figure 6. Applications of nanotechnology in precision agriculture.

FUTURE SCENARIOS

Nanotechnology is a capable domain of recent times. Metallic nanoparticles are mainly produced during enzymatic reactions. Mechanistic understandings may provide control over morphology, stability, and rate of synthesis of nanomaterials. Environmental contaminants are remediated by biologically synthesized nanomaterials (Masood, et al. 2013). Heavy metal pollution has been seen as a significant concern because of its hazardous impacts and non-biodegradable nature (Wu, et al. 2016). Bioremediation of heavy metal pollution has been preferred over other methods such as chemical precipitation and electrochemical methods (Diaz, et al. 2015). Microbes have been evolved and can offer inventive bio-based clean-up methods (Engel, 2017). Metal reducing microbes promotes precipitations, minerals dissolutions and transformations such as geochemical agents (Neumann, 2017). Microbial reduction methods have delivered a new way to explore nanotechnology. Bio-mineralization is among, one of them which controls the environmental impact of heavy metals (Kang, et al. 2017). The enzyme metal interaction here provides detoxification of heavy metals by a biotransformation process. Microbes that produce nitrate reductase can enhance the efficacy of the bioremediation process. Other enzymes along with nitrate reductase have been studied for further possibilities of heavy metal remediation. Microbe metal interaction and the importance of their enzymes still required further investigation (Waseem, et al. 2019). The studies of these microbial interactions at the genetic level may provide new genetic tools for bioremediation. These findings may assist in clarifying new phenomena such as antibiotic resistance (Ghosh, et al. 2012). The upsurge in antibiotic resistance is becoming a serious concern all over the world for modern medicine. The combination of metallic nanomaterials with antibiotics may overcome these problems such as bacterial resistance (Jang, et al. 2018). However, the co-occurrence of antibiotic resistance and metallic resistance genes has been reported recently. Thus microbes may play a critical role in controlling heavy metal pollution. The use of microbial nanomaterials for renewable energy needs further investigation.

CONCLUSION

Development in engineering and nanoscience technology may prove more helpful for generating new chances to develop more cost-effective and environmentally safe agricultural technologies. Nanoparticle has peculiar properties which make them more suitable to solve pollution problems. Current findings have indicated that nanomaterials are beneficial particles for heavy metal removal because of their unique and efficient properties in removing heavy metals from the agricultural system. The microbial threats to agricultural quality and production may also be removed with the help of nanotechnology. Zinc, chitosan and copper-based nanomaterials possess antimicrobial and antifungal properties which in turn can solve these microbial problems in agriculture and may increase agricultural production. Thus further research must be made in this direction to develop nanoparticles with greater stability, specificity and capability for simultaneous removal of multiple contaminants, and microbial pathogens from

agriculture. It is not clear that why and how different species of plants show dissimilar resistance properties to nanoparticles. It is recommended to discover the effects of environmental factors on the uptake and accumulation of nanoparticles by different plant species. The route of nanoparticles movement in plant's cells and their effect on the genetic system of plants should be investigated.

CONFLICT OF INTERESTS

We do not corroborate that this work is original and has not been published elsewhere nor is it currently under consideration for publication elsewhere. We have no conflicts of interest concerning the publication of this paper.

REFERENCES

1. Ahmad N, Sharma S, Singh VN, Shamsi SF, Fatma A, Mehta BR. (2011). Biosynthesis of silver nanoparticles from *Desmodium triflorum*: A novel approach towards weed utilization. *Biotechnol Res Int*. 2011.
2. Ahmaruzzaman M (2011). Industrial wastes as low-cost potential adsorbents for the treatment of wastewater laden with heavy metals. *Adv colloid interface sci*. 166(1-2): 36-59.
3. Ahmed E, Kalathil S, Shi L, Alharbi O, Wang P (2018). Synthesis of ultra-small platinum, palladium and gold nanoparticles by *Shewanella loihica* PV-4 electrochemically active biofilms and their enhanced catalytic activities. *J Saudi Chem Soci*. 22(8): 919-929.
4. Ankamwar B (2010). Biosynthesis of gold nanoparticles (green-gold) using leaf extract of *Terminalia catappa*. *J Chem*. 7(4): 1334-1339.
5. Ankamwar B, Chaudhary M, Sastry M. (2005). Gold nano triangles biologically synthesized using tamarind leaf extract and potential application in vapour sensing. *Synthesis and Reactivity in Inorganic, Metal-Organic and Nano-Metal Chemistry*. 35(1): 19-26.
6. Arfin T, Yadav N (2013). Impedance characteristics and electrical double-layer capacitance of composite polystyrene-cobalt-arsenate membrane. *J Ind Eng Chem*. 19(1): 256-262.
7. Aromal SA, Vidhu VK, Philip D. (2012). Green synthesis of well-dispersed gold nanoparticles using *Macrotyloma uniflorum*. *Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy*. 85(1): 99-104.
8. Arulkumar S, Sabesan M (2010). Biosynthesis and characterization of gold nanoparticles using antiparkinsonian drug *Mucuna pruriens* plant extract. *Int Res Pharm Sci*. 1: 417-20.
9. Babu MG, Gunasekaran P (2009). Production and structural characterization of crystalline silver nanoparticles from *Bacillus cereus* isolate. *Colloids surf B Biointerfaces*. 74(1): 191-195.

10. Bali R, Razak N, Lumb A, Harris AT. (2006) July. The synthesis of metallic nanoparticles inside live plants. In Nanoscience and Nanotechnology. 2006. ICONN'06. International Conference on. IEEE.
11. Bankar A, Joshi B, Kumar AR, Zinjarde S (2010). Banana peel extract mediated synthesis of gold nanoparticles. Colloids Surfaces B Biointerfaces. 80(1): 45-50.
12. Bhati-Kushwaha H, Malik CP (2012). Biomimetic of Silver Nanoparticles from the dried extract of *Tridax procumbens* L. (Stem and Leaf). J Plant Sci Res, 28(2).
13. Blaney LM, Cinar S, SenGupta AK (2007). Hybrid anion exchanger for trace phosphate removal from water and wastewater. Water res. 41(7): 1603-1613.
14. Boruah SK, Boruah PK, Sarma P, Medhi C, Medhi OK (2012). Green synthesis of gold nanoparticles using *Camellia sinensis* and kinetics of the reaction. Adv Mater Lett, 3(6): 481-486.
15. Chandran SP, Chaudhary M, Pasricha R, Ahmad A. Sastry M (2006). Synthesis of gold nano triangles and silver nanoparticles using *Aloe vera* plant extract. Biotechnology progr. 22(2): 577-583.
16. Chen X, Zhang W, Luo X, Zhao F, Li Y, Li R, Li Z (2017). Efficient removal and environmentally benign detoxification of Cr (VI) in aqueous solutions by Zr (IV) cross-linking chitosan magnetic microspheres. Chemosphere. 185: 991-1000.
17. Dhas SP, Mukerjee AMITAVA, Chandrasekaran N (2013). Phytosynthesis of silver nanoparticles using *Cerops tagal* and its antimicrobial potential against human pathogens. Int J Pharm Pharm Sci. 5(3): 349-352.
18. Diaz MR, Swart PK, Eberli GP, Oehlert AM, Devlin Q, Saeid A, Altabet MA et al (2015). Geochemical evidence of microbial activity within ooids. Sedimentology. 62(7): 2090-2112.
19. Duhan JS, Kumar R, Kumar N, Kaur P, Nehra K, Duhan S (2017). Nanotechnology: The new perspective in precision agriculture. Biotechnol Rep. 15: 11-23.
20. Dwivedi AD, Gopal K (2011). Plant-mediated biosynthesis of silver and gold nanoparticles. Journal of biomedical nanotechnology. 7(1): 163-164.
21. Engel J (2017). Biominerals and Their Function in Different Organisms. In A Critical Survey of Biomineralization. Springer Cham. 7-11.
22. Ewecharoen A, Thiravetyan P, Wendel E, Bertagnolli H (2009). Nickel adsorption by sodium polyacrylate-grafted activated carbon. J Hazard Mater. 171(1-3): 335-339.
23. Farooqui MA, Chauhan PS, Krishnamoorthy P, Shaik J (2010). Extraction of silver nanoparticles from the leaf extracts of *Clerodendrum inerme*. Dig J Nanomater Biostruct. 5(1): 43-49.
24. Fazaludeena MF, Manickamb C, Ashanky IM, Ahmed MQ. Bege QZ (2017). Synthesis and characterizations of gold nanoparticles by *Justicia gendarussa* Burm F leaf extract. J Microbiol Biotechnol Res. 2(1): 23-34.
25. Fu L, Wang Z, Dhankher OP. Xing B (2020). Nanotechnology as a new sustainable approach for controlling crop diseases and increasing agricultural production. J Exp Bot. 71(2): 507-519.
26. Gardea-Torresdey JL, Parsons JG, Gomez E, Peralta-Videa J, Troiani HE, Santiago P, Yacaman MJ et al. (2002). Formation and growth of Au nanoparticles inside live alfalfa plants. Nano letters. 2(4): 397-401.
27. Gericke M. Pinches A (2006). Biological synthesis of metal nanoparticles. Hydrometallurgy. 83(1-4): 132-140.
28. Ghosh S, Patil S, Ahire M, Kitture R, Kale S, Pardesi K, Cameotra SS et al (2012). Synthesis of silver nanoparticles using *Dioscorea bulbifera* tuber extract and evaluation of its synergistic potential in combination with antimicrobial agents. Int J Nanomed. 7: 483-496.
29. Gnanadesigan M, Anand M, Ravikumar S, Maruthupandy M, Ali MS, Vijayakumar V, Kumaraguru AK et al (2012). Antibacterial potential of biosynthesised silver nanoparticles using *Avicennia marina* mangrove plant. Appl Nanosci. 2(2): 143-147.
30. Gnanajobitha G, Annadurai G. Kannan C (2012). Green synthesis of silver nanoparticle using *Elettaria cardamom* and assesment of its antimicrobial activity. Int. J Pharma Sci Res (IJPSR). 3: 323-330.
31. Gudkov SV, Shafeev GA, Glinushkin AP, Shkirin AV, Barmina EV, Rakov II, Simakin AV et al (2020). Production and Use of Selenium Nanoparticles as Fertilizers. ACS omega. 5(8): 17767-17774.
32. Guha T, Gopal G, Kundu R. Mukherjee A (2020). Nanocomposites for delivering agrochemicals: A comprehensive review. J Agric Food Chem. 68(12): 3691-3702.
33. Gurunathan S, Kalishwaralal K, Vaidyanathan R, Venkataraman D, Pandian SRK, Muniyandi J, Hariharan N et al (2009). Biosynthesis, purification and characterization of silver nanoparticles using *Escherichia coli*. Colloids Surf B Biointerfaces. 74(1): 328-335.
34. He S, Zhang Y, Guo Z, Gu N (2008). Biological synthesis of gold nanowires using extract of *Rhodospseudomonas capsulata*. Biotechnol prog. 24(2): 476-480.
35. Hočevár M, Krašovec UO, Bokalič M, Topič M, Veurman W, Brandt H, Hinsch A (2013). Sol-gel based TiO₂ paste applied in screen-printed dye-sensitized solar cells and modules. J Ind Eng Chem. 19(5): 1464-1469.

36. Hong SH, Kwon SN, Bae JS, Song MY (2012). Hydrogen storage characteristics of melt spun Mg–23.5 Ni–5Cu alloys mixed with LaNi₅ and/or Nb₂O₅. *J Ind Eng Chem.*18(1): 61-64.
37. Hsu YS, Lou JC, Chou MS, Hsu KL, Han JY (2015). Characterization of single-walled carbon nanotubes and adsorption of perchlorate in water. *Water, Air, & Soil Pollution.* 226(3): 58.
38. Hussein MI, El-Aziz MA, Badr Y, Mahmoud MA (2007). Biosynthesis of gold nanoparticles using *Pseudomonas aeruginosa*. *Spectrochim Acta A Mol Biomol Spectrosc.* 67(3-4): 1003-1006.
39. Ingale AG, Chaudhari AN (2013). Biogenic synthesis of nanoparticles and potential applications: an eco-friendly approach. *J Nanomed Nanotechnol.* 4(165): 1-7.
40. Iqbal MA (2019). Nano-Fertilizers for Sustainable Crop Production under Changing Climate: A Global Perspective. In *Sustainable Crop Production*. IntechOpen.
41. Jain D, Daima HK, Kachhwaha S, Kothari SL (2009). Synthesis of plant-mediated silver nanoparticles using papaya fruit extract and evaluation of their anti-microbial activities. *Dig J Nanomater Biostruct.* 4(3): 557-563.
42. Jain D, Kachhwaha S, Jain R, Srivastava G, Kothari SL (2010). Novel microbial route to synthesize silver nanoparticles using spore crystal mixture of *Bacillus thuringiensis*. *Indian J Expo Biol.* 48(11): 1152-1156.
43. Jain PK, Lee KS, El-Sayed IH, El-Sayed MA (2006). Calculated absorption and scattering properties of gold nanoparticles of different size, shape, and composition: applications in biological imaging and biomedicine. *J Phys Chem B.* 110(14): 7238-7248.
44. Jampilek J, Kráľová K, (2015). Application of nanotechnology in agriculture and food industry, its prospects and risks. *Ecol Chem Eng S.* 22(3): 321-361.
45. Jang HM, Lee J, Kim YB, Jeon JH, Shin J, Park MR, Kim YM (2018). Fate of antibiotic resistance genes and metal resistance genes during thermophilic aerobic digestion of sewage sludge. *Bioresour Technol.* 249: 635-643.
46. Joglekar S, Kodam K, Dhaygude M, Hudlikar M (2011). Novel route for rapid biosynthesis of lead nanoparticles using aqueous extract of *Jatropha curcas* L. latex. *Mater Lett.* 65(19-20): 3170-3172.
47. Juibari MM, Abbasalizadeh S, Jouzani GS, Noruzi M (2011). Intensified biosynthesis of silver nanoparticles using a native extremophilic *Ureibacillus thermosphaericus* strain. *Mater Lett.* 65(6): 1014-1017.
48. Jung JY, Lee SH, Kim JM, Park MS, Bae JW, Hahn Y, Madsen EL et al (2011). Metagenomic analysis of kimchi, a traditional Korean fermented food. *Appl Environ Microbiol.* 77(7): 2264-2274.
49. Kalavathy H (2010). Studies on heavy metal removal from aqueous solutions and effluent.
50. Kalia A, Sharma SP, Kaur H, Kaur H (2020). Novel nanocomposite-based controlled-release fertilizer and pesticide formulations: prospects and challenges. In *Multifunctional Hybrid Nanomaterials for Sustainable Agri-Food and Ecosystems*. Elsevier. 99-134.
51. Kang F, Qu X, Alvarez PJ, Zhu D (2017). Extracellular saccharide-mediated reduction of Au³⁺ to gold nanoparticles: new insights for heavy metals biomineralization on microbial surfaces. *Environ sci technol.* 51(5): 2776-2785.
52. Kasthuri J, Veerapandian S, Rajendiran N. (2009). Biological synthesis of silver and gold nanoparticles using apiin as reducing agent. *Colloids Surf B Biointerfaces.* 68(1): 55-60.
53. Kathad U, Gajera HP (2014). Synthesis of copper nanoparticles by two different methods and size comparison. *Int J Pharm Bio Sci.* 5(3): 533-540.
54. Kavitha KS, Baker S, Rakshith D, Kavitha HU, Yashwantha Rao HC, Harini BP, Satish S (2013). Plants are a green source for the synthesis of nanoparticles. *Int Res J Biol Sci.* 2(6): 66-76.
55. Kim JH, Hong YJ, Park BK, Kang YC (2013). Nano-sized LiNi_{0.5}Mn_{1.5}O₄ cathode powders with good electrochemical properties prepared by high temperature flame spray pyrolysis. *J Ind Engin Chem.* 19(4): 1204-1208.
56. Klaus-Joerger T, Joerger R, Olsson E, Granqvist CG (2001). Bacteria as workers in the living factory: metal-accumulating bacteria and their potential for materials science. *TRENDS in Biotechnol.* 19(1): 15-20.
57. Lee JH, Hwang ET, Kim BC, Lee SM, Sang BI, Choi YS, Kim J et al (2007). Stable and continuous long-term enzymatic reaction using an enzyme–nanofiber composite. *Appl microbiol biotechnol.* 75(6): 1301-1307.
58. Lee JS, Shin DH, Jun J, Lee C, Jang J (2014). Fe₃O₄/Carbon Hybrid Nanoparticle Electrodes for High-Capacity Electrochemical Capacitors. *ChemSusChem.* 7(6): 1676-1683.
59. Li H, Li W, Zhang Y, Wang T, Wang B, Xu W, Jiang L et al (2011). Chrysanthemum-like α -FeOOH microspheres produced by a simple green method and their outstanding ability in heavy metal ion removal. *J Mater Chem.* 21(22): 7878-7881.
60. Li L, Tang Q, Li H, Yang X, Hu W, Song Y, Shuai Z et al (2007). An ultra-closely π -stacked organic semiconductor for high-performance field-effect transistors. *Adv mater.* 19(18): 2613-2617.

61. Li S, Anderson TA, Maul JD, Shrestha B, Green MJ, Cañas-Carrell JE (2013). Comparative studies of multi-walled carbon nanotubes (MWNTs) and octadecyl (C18) as sorbents in passive sampling devices for biomimetic uptake of polycyclic aromatic hydrocarbons (PAHs) from soils. *Science of the Total Environment*. 461: 560-567.
62. Li S, Shen Y, Xie A, Yu X, Zhang X, Yang L, Li C (2007). Rapid, room-temperature synthesis of amorphous selenium/protein composites using *Capsicum annum* L. extract. *Nanotechnology*. 18(40): 405101.
63. Li XQ, Zhang WX (2007). Sequestration of metal cations with zerovalent iron nanoparticles a study with high resolution X-ray photoelectron spectroscopy (HR-XPS). *J Phy Chem C*. 111(19): 6939-6946.
64. Lin D, Xing B (2007). Phytotoxicity of nanoparticles: inhibition of seed germination and root growth. *Environ Pollut*. 150(2): 243-250.
65. Lu C, Chiu H (2006). Adsorption of zinc (II) from water with purified carbon nanotubes. *Chem Eng Sci*. 61(4): 1138-1145.
66. Mallikarjuna K, Narasimha G, Dillip GR, Praveen B, Shreedhar B, Lakshmi CS, Reddy BVS et al (2011). Green synthesis of silver nanoparticles using *Ocimum* leaf extract and their characterization. *Dig J Nanomater Biostruct*. 6(1): 181-186.
67. Mandal D, Bolander ME, Mukhopadhyay D, Sarkar G, Mukherjee P (2006). The use of microorganisms for the formation of metal nanoparticles and their application. *Appl microbial biotechnol*. 69(5): 485-492.
68. Masood F, Malik A (2013). Current Aspects of Metal Resistant Bacteria in Bioremediation: From Genes to Ecosystem. In *Management of Microbial Resources in the Environment*. Springer, Dordrecht: 289-311.
69. Mathew L, Chandrasekaran N, Raichur AM. Mukherjee A (2010). Biomimetic synthesis of nanoparticles: science, technology and applicability. *Biomimetics: Learning from Nature*.
70. Monica RC, Cremonini R (2009). Nanoparticles and higher plants. *Caryologia*. 62(2): 161-165.
71. MubarakAli D, Thajuddin N, Jeganathan K, Gunasekaran M (2011). Plant extract mediated synthesis of silver and gold nanoparticles and its antibacterial activity against clinically isolated pathogens. *Colloids Surf B Biointerfaces*. 85(2): 360-365.
72. Mueller NC, Braun J, Bruns J, Černík M, Rissing P, Rickerby D, Nowack B (2012). Application of nanoscale zero valent iron (NZVI) for groundwater remediation in Europe. *Environ Sci Pollut Res int*. 19(2): 550-558.
73. Muhammad Z, Inayat N, Majeed A. (2020). Application of Nanoparticles in Agriculture as Fertilizers and Pesticides: Challenges and Opportunities. In *New Frontiers in Stress Management for Durable Agriculture* Springer. Singapore: 281-293.
74. Murugadoss G, Rajamannan B, Madhusudhanana U. (2009). Synthesis and characterization of water-soluble ZnS: Mn²⁺ nanocrystals. *Chalcogenide Lett*. 6(5).
75. Nair B, Pradeep T (2002). Coalescence of nanoclusters and formation of submicron crystallites assisted by *Lactobacillus* strains. *Cryst Growth Des*. 2(4): 293-298.
76. Nanda A, Saravanan M (2009). Biosynthesis of silver nanoparticles from *Staphylococcus aureus* and its antimicrobial activity against MRSA and MRSE. *Nanomedicine: Nanotechnology, Biology and Medicine*. 5(4): 452-456.
77. Narayanan KB, Sakthivel N (2008). Coriander leaf mediated biosynthesis of gold nanoparticles. *Mater Lett*. 62(30): 4588-4590.
78. Nazeruddin GM, Prasad SR, Shaikh YI, Ansari J, Sonawane KD, Nayak AK, Deshmukh MB et al (2016). In-vitro Bio-fabrication of Multi-applicative Silver Nanoparticles using *Nicotiana tabacum* leaf extract. *Res J Life Sci Bioinformat Pharmaceut Chem Sci*. 2:6-33.
79. Ndlovu N, Mayaya T, Muitire C, Munyengwa N (2020). Nanotechnology Applications in Crop Production and Food Systems. *Int J Plant Breed*. 7(1): 624-634.
80. Neumann W, Gulati A, Nolan EM. (2017). Metal homeostasis in infectious disease: recent advances in bacterial metallophores and the human metal-withholding response. *Curr Opin Chem Biol*. 37:10-18.
81. Olenin AY, Lisichkin GV (2011). Metal nanoparticles in condensed media: preparation and the bulk and surface structural dynamics. *Russian Chemical Reviews*. 80(7): 605-630.
82. Pandey G (2020). Agri-Nanotechnology for Sustainable Agriculture. In *Ecological and Practical Applications for Sustainable Agriculture* Springer, Singapore: 229-249.
83. Parashar V, Parashar R, Sharma B, Pandey AC (2009). Parthenium leaf extract mediated synthesis of silver nanoparticles: a novel approach towards weed utilization. *Dig J Nanomater Biostruct (DJNB)*. 4(1): 45-50.
84. Parida UK, Bindhani BK, Nayak P (2011). Green synthesis and characterization of gold nanoparticles using onion (*Allium cepa*) extract. *World Journal of Nano Science and Engineering*. 1(04): 93-98.
85. Park SJ, Kim YM (2005). Adsorption behaviors of heavy metal ions onto electrochemically oxidized activated carbon fibers. *Mat Sci Eng A*. 391(1-2): 121-123.
86. Pavani KV, Swati T, Snehik V, Sravya K, Sirisha M (2012). Phytofabrication of lead nanoparticles using Grape skin extract. *Int J Eng Sci Technol*. 4(7).

87. Pejman A, Bidhendi GN, Ardestani M, Saeedi M, Baghvand A (2015). A new index for assessing heavy metals contamination in sediments: a case study. *Ecological indicators*. 58: 365-373.
88. Petla RK, Vivekanandhan S, Misra M, Mohanty AK, Satyanarayana N (2012). Soybean (*Glycine max*) leaf extract based green synthesis of palladium nanoparticles. *J Biomater Nanobiotechnol*. 3(1): 14-19.
89. Philip D (2009). Biosynthesis of Au, Ag and Au–Ag nanoparticles using edible mushroom extract. *Spectrochim Acta A Mol Biomol Spectrosc*. 73(2): 374-381.
90. Philip D (2010). Green synthesis of gold and silver nanoparticles using *Hibiscus rosa sinensis*. *Physica E Low Dimens Syst Nanostruct*. 42(5): 1417-1424.
91. Pope JH, Aufderheide TP, Ruthazer R, Woolard RH, Feldman JA, Beshansky JR, Griffith JL et al (2000). Missed diagnoses of acute cardiac ischemia in the emergency department. *N Engl J Med*. 342(16): 1163-1170.
92. Popescu M, Velea A, Lőrinczi A. (2010). Biogenic Production of Nanoparticles. *Dig J Nanomater Biostruct (DJNB)*. 5(4).
93. Prabhu S, Poulose EK (2012). Silver nanoparticles: mechanism of antimicrobial action, synthesis, medical applications, and toxicity effects. *Int nano lett*. 2(1): 32.
94. Priya MM, Selvi BK, Paul JA (2011). Green synthesis of silver nanoparticles from the leaf extracts of *Euphorbia hirta* and *Nerium indicum*. *Dig J Nanomater Biostruct (DJNB)*. 6(2): 869-877.
95. Pugazhenthiran N, Anandan S, Kathiravan G, Prakash NKU, Crawford S, Ashokkumar M (2009). Microbial synthesis of silver nanoparticles by *Bacillus* sp. *J Nanoparticle Res*. 11(7): 1811-1815.
96. Raghunandan D, Bedre MD, Basavaraja S, Sawle B, Manjunath SY, Venkataraman A (2010). Rapid biosynthesis of irregularly shaped gold nanoparticles from macerated aqueous extracellular dried clove buds (*Syzygium aromaticum*) solution. *Colloids Surf B Biointerfaces*. 79(1): 235-240.
97. Rao GP, Lu C, Su F (2007). Sorption of divalent metal ions from aqueous solution by carbon nanotubes: a review. *Sep Purif Technol*. 58(1): 224-231.
98. Rathnayaka RMNN, Iqbal YB, Rifnas LM (2018). Influence of urea and nano-nitrogen fertilizers on the growth and yield of rice (*Oryza sativa* L.) cultivar 'Bg 250'. Influence of Urea and Nano-Nitrogen Fertilizers on the Growth and Yield of Rice (*Oryza sativa* L.) Cultivar 'Bg 250'. 5(2):7-8.
99. Rathore KS, Patidar D, Janu Y, Saxena NS, Sharma K, Sharma TP (2008). Structural and optical characterization of chemically synthesized ZnS nanoparticles. *Chalcogenide Letters*. 5(6): 105-110.
100. Raut Rajesh W, Lakkakula Jaya R, Kolekar Niranjana S, Mendhulkar Vijay D, Kashid Sahebrao B (2009). Photosynthesis of silver nanoparticles using *Gliricidia sepium* (Jacq.). *Curr Nanosci*. 5(1): 117-122.
101. Raut RW, Kolekar NS, Lakkakula JR, Mendhulkar VD, Kashid SB (2010). Extracellular synthesis of silver nanoparticles using dried leaves of *Pongamia pinnata* (L) pierre. *Nano-Micro Letters*. 2(2): 106-113.
102. Reddy MC, Murthy KR, Srilakshmi A, Rao KS, Pullaiah T (2015). Photosynthesis of eco-friendly silver nanoparticles and biological applications—a novel concept in nanobiotechnology. *Afr J Biotechnol*. 14(3): 222-247.
103. Rehmanullah ZM, Inayat N, Majeed A (2020). Application of Nanoparticles in Agriculture as Fertilizers. *New Frontiers in Stress Management for Durable Agriculture*. 281.
104. Robinson T, McMullan G, Marchant R, Nigam P (2001). Remediation of dyes in textile effluent: a critical review on current treatment technologies with a proposed alternative. *Bioresour technol*. 77(3): 247-255.
105. Rodríguez-León E, Iñiguez-Palomares R, Navarro RE, Herrera-Urbina R, Tánori J, Iñiguez-Palomares C, Maldonado A (2013). Synthesis of silver nanoparticles using reducing agents obtained from natural sources (*Rumex hymenosepalus* extracts). *Nanoscale Res Lett*. 8(1): 318.
106. Roychoudhury A (2020). Silicon-Nanoparticles in Crop Improvement and Agriculture. *Int J Rec Adv Biotechnol Nanotechnol*. 3(1).
107. Russell D (2016). AEMS Compendium. U.S. Patent Application 14/544,443.
108. Sathishkumar M, Sneha K, Kwak IS, Mao J, Tripathy SJ, Yun YS (2009). Phyto-crystallization of palladium through reduction process using *Cinnamom zeylanicum* bark extract. *J Hazard Mater*. 171(1-3): 400-404.
109. Sathyavathi R, Krishna M.B, Rao SV, Saritha R, Rao DN (2010). Biosynthesis of silver nanoparticles using *Coriandrum sativum* leaf extract and their application in nonlinear optics. *Advanced science letters*. 3(2): 138-143.
110. Saxena J, Sharma MM, Gupta S, Singh A (2014). The emerging role of fungi in nanoparticle synthesis and their applications. *World J Pharm Sci*. 3:1586-1613.
111. Sekhon BS (2010). Food nanotechnology—an overview. *Nanotechnology sci appl*. 3:1-15.
112. Shankar SS, Ahmad A, Pasricha R, Sastry M (2003). Bio-reduction of chloroaurate ions by geranium leaves and its endophytic fungus yields gold nanoparticles of different shapes. *J Mater Chem*. 13(7): 1822-1826.

113. Shebl A, Hassan AA, Salama DM, Abd El-Aziz ME, Abd Elwahed MS (2020). Template-free microwave-assisted hydrothermal synthesis of manganese zinc ferrite as a nano fertilizer for squash plant (*Cucurbita pepo* L). *Heliyon*. 6(3): e03596.
114. Shebl A, Hassan AA, Salama DM, El-Aziz A, Abd Elwahed MS (2019). Green synthesis of nano fertilizers and their application as a foliar for *Cucurbita pepo* L. *Journal of Nanomaterials*. 2019.
115. Shukla YM (2019). Nanofertilizers: A Recent Approach in Crop Production. In *Nanotechnology for Agriculture: Crop Production & Protection*. Singapore. 25-58.
116. Siddiqui MH, Al-Wahaibi MH, Mohammad F eds (2015). *Nanotechnology and plant sciences: nanoparticles and their impact on plants*. Springer.
117. Singh SK, Kasana RC, Yadav RS, Pathak R (2020). Current Status of Biologically Produced Nanoparticles in Agriculture. In *Biogenic Nano-Particles and their Use in Agro-ecosystems*. 393-406.
118. Sintubin L, De Windt W, Dick J, Mast J, van der Ha D, Verstraete W, Boon N (2009). Lactic acid bacteria as reducing and capping agent for the fast and efficient production of silver nanoparticles. *Appl Microbiol Biotechnol*. 84(4): 741-749.
119. Smitha SL, Philip D, Gopchandran KG (2009). Green synthesis of gold nanoparticles using *Cinnamomum zeylanicum* leaf broth. *Spectrochim Acta A Mol Biomol Spectrosc*. 74(3): 735-739.
120. Srivastava NK, Majumder CB (2008). Novel biofiltration methods for the treatment of heavy metals from industrial wastewater. *J hazard mater* 151(1): 1-8.
121. Sweeney RY, Mao C, Gao X, Burt JL, Belcher AM, Georgiou G, Iverson BL (2004). Bacterial biosynthesis of cadmium sulfide nanocrystals. *Chem boil*. 11(11): 1553-1559.
122. Tang WW, Zeng GM, Gong JL, Liu Y, Wang XY, Liu YY, Liu ZF et al (2012). Simultaneous adsorption of atrazine and Cu (II) from wastewater by magnetic multi-walled carbon nanotube. *Chem Eng J*. 211: 470-478.
123. Turhan T, Avcıbası YG, Sahiner N (2013). Versatile p (3-sulfopropyl methacrylate) hydrogel reactor for the preparation of Co, Ni nanoparticles and their use in hydrogen production. *J Industria Eng Chem*. 19(4): 1218-1225.
124. Umashankari J, Inbakandan D, Ajithkumar TT, Balasubramanian T (2012). Mangrove plant, *Rhizophora mucronata* (Lamk, 1804) mediated one-pot green synthesis of silver nanoparticles and its antibacterial activity against aquatic pathogens. *Aquat biosyst*. 8(1): 11.
125. Usman M, Farooq M, Wakeel A, Nawaz A, Cheema SA, ur Rehman H, Ashraf I et al (2020). *Nanotechnology in agriculture: Current status, challenges and future opportunities*. *Science of The Total Environment*. 137778.
126. Vahabi K, Mansoori GA, Karimi S (2011). Biosynthesis of silver nanoparticles by fungus *Trichoderma reesei* (a route for large-scale production of AgNPs). *Insciences J*. 1(1): 65-79.
127. Vankar PS, Bajpai D (2010). Preparation of gold nanoparticles from *Mirabilis jalapa* flowers.
128. Velammal SP, Devi TA, Amaladhas TP (2016). Antioxidant, antimicrobial and cytotoxic activities of silver and gold nanoparticles synthesized using *Plumbago zeylanica* bark. *J Nanostructure Chem*. 6(3): 247-260.
129. Wang P, Lombi E, Zhao FJ, Kopittke PM (2016). Nanotechnology: a new opportunity in plant sciences. *Trends in plant science*. 21(8): 699-712.
130. Wang YS, Shan XQ, Feng MH, Chen GC, Pei ZG, Wen B, Liu T (2009). Effects of copper, lead, and cadmium on the sorption of 2, 4, 6-trichlorophenol onto and desorption from wheat ash and two commercial humic acids. *Environ sci technol*. 43(15): 5726-5731.
131. Waseem H, Jameel S, Ali J, Saleem Ur Rehman H, Tauseef I, Farooq U, Jamal A et al (2019). Contributions and challenges of high throughput qPCR for determining antimicrobial resistance in the environment: a critical review. *Molecule*. 24(1):163.
132. Wiyaratn W, Appamana W, Charojrochkul S, Kaewkuekool S, Assabumrungrat S (2012). Au/La1-xSrxMnO3 nanocomposite for chemical-energy cogeneration in solid oxide fuel cell reactor. *J Indust Eng Chem*. 18(5): 1819-1823.
133. Wu W, Huang H, Ling Z, Yu Z, Jiang Y, Liu P, Li X (2016). Genome sequencing reveals mechanisms for heavy metal resistance and polycyclic aromatic hydrocarbon degradation in *Delftia lacustris* strain LZ-C. *Ecotoxicology*. 25(1): 234-247.
134. Yadav KK, Singh JK, Gupta N, Kumar V (2017). A review of nanobioremediation technologies for environmental cleanup: A novel biological approach. *J Mater Environ. Sci*. 8: 740-757.
135. Yang X, Li Q, Wang H, Huang J, Lin L, Wang W, Sun D (2010). Green synthesis of palladium nanoparticles using broth of *Cinnamomum camphora* leaf. *J Nanoparticle Res*. 12(5): 1589-1598.
136. Zhang, WX (2003). Nanoscale iron particles for environmental remediation: an overview. *J nanoparticle Res*. 5(3-4): 323-332.

137. Zhong HX, Ma YL, Cao XF, Chen XT, Xue ZL (2009). Preparation and characterization of flowerlike $\text{Y}_2(\text{OH})_5\text{NO}_3 \cdot 1.5 \text{H}_2\text{O}$ and Y_2O_3 and their efficient removal of Cr (VI) from aqueous solution. *The J Phys Chem C*. 113(9): 3461-3466.
138. Zhong LS, Hu JS, Cao AM, Liu Q, Song WG, Wan LJ (2007). 3D flowerlike ceria micro/nanocomposite structure and its application for water treatment and CO removal. *Chem Mater*. 19(7): 1648-1655.
139. Zhu B, Wu S, Xia X, Lu X, Zhang X, Xia N, Liu T (2016). Effects of carbonaceous materials on microbial bioavailability of 2, 2', 4, 4'-tetrabromodiphenyl ether (BDE-47) in sediments. *J hazard mater* 312: 216-223.